## SNC 2P - 2020-06-09 (Tuesday)

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Please Check In: <u>https://forms.office.com/Pages/ResponsePage.aspx?id=GAmpRLReCU2WCd35</u> yhGvQuASPjs6aYVFk-EAh60FvohURDBEQ1RKTDFSREFRQTZZNktCOUIFRDU1RCQIQCN0P Wcu

Hello All:

Finally ready. I have created another youtube video to help through the next bit of nomenclature.

The goal today is to:

Watch this video more than once if necessary: <u>https://www.youtube.com/watch?</u> <u>v=Qd68HtyIPFc&feature=youtu.be</u>

Look at this periodic table and understand why some of the boxes are colour green: <u>http://www.s</u> <u>chlenkerchem.org/2P/Chemistry/periodic%20tables/periodic%20tabl%20showing%20multiple%</u> <u>20oxidation%20states.pdf</u>

Look at and understand three examples: <u>http://www.schlenkerchem.org/2P/Chemistry/Notes/not</u> <u>e%205%20-%20nomenclature%20polyvalent%20examples.pdf</u>

Work on the this worksheet to complete the following selection of questions:

http://www.schlenkerchem.org/2P/Chemistry/Worksheets/ws%20nomenclature%203%20ionic %20compounds%20grouped%20and%20old%20sheet.pdf

Question #1: a to e and k to o only (I want to save the remaining question for tomorrow)

Question #2: a to e and k to o only (I want to save the remaining question for tomorrow)

Use answers if you need to: <u>http://www.schlenkerchem.org/2P/Chemistry/Worksheets/ws%20no</u> menclature%203%20ionic%20compounds%20grouped%20and%20old%20sheet%20answers.p <u>df</u>

Please do the best you can. It turns out that the ability to think this way about formula is critical for future study in chemistry. You can trust me on this one. I have been a chemistry teacher for 28 years!

## Mr. S.

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