

SNC 2P - 2020-06-02 (Tuesday)

From: Fred Schlenker <fred_schlenker@bwdsb.on.ca>

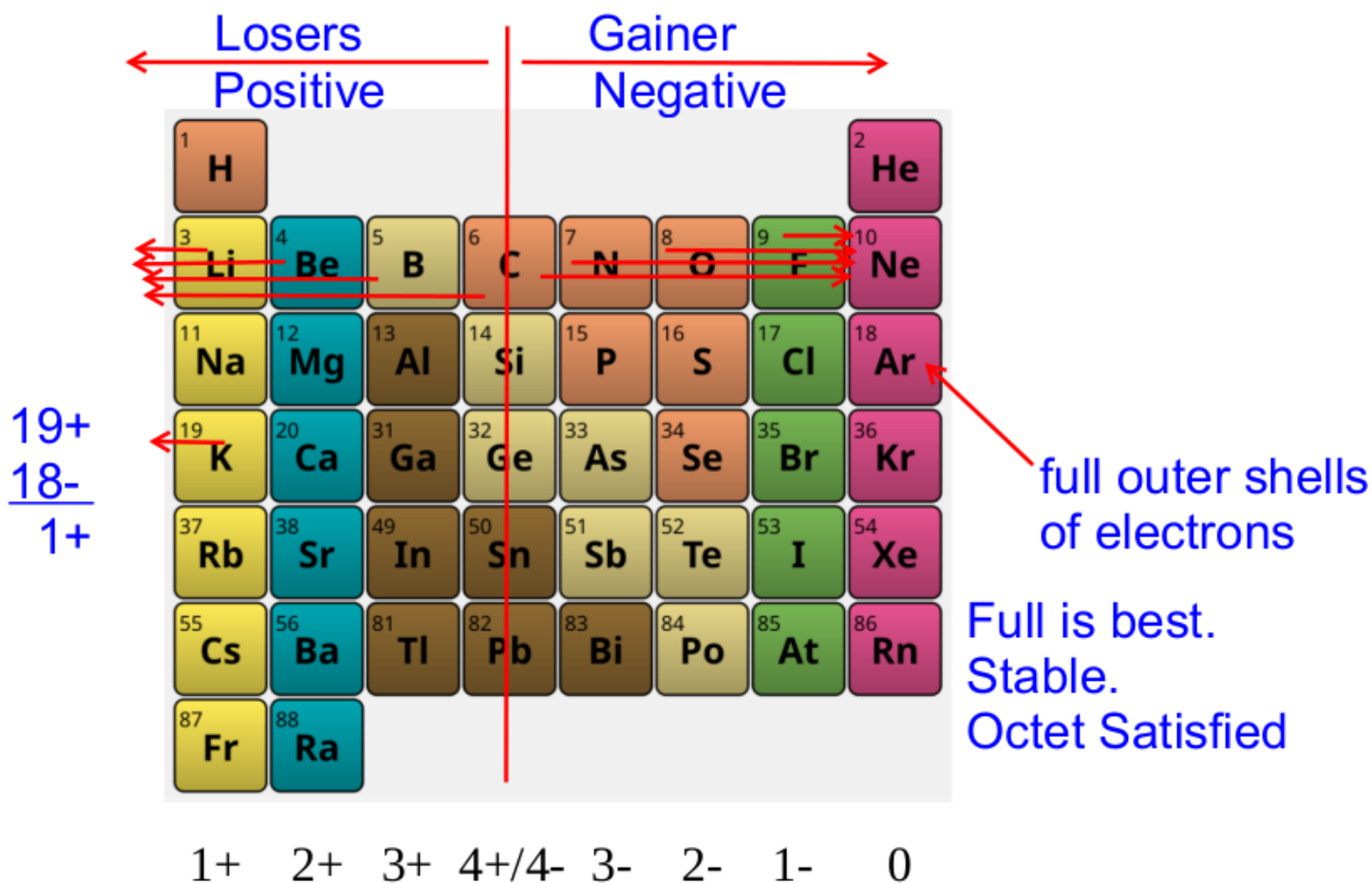
Date: Tue, 2 Jun 2020 13:19:57 +0000 (2020-06-02 09:19:57 AM)

Good Morning All:

Please check in: <https://forms.office.com/Pages/ResponsePage.aspx?id=G&id=GAmpRLReCU2WCd35yhGvQuASPjs6aYVfK-EAh60FvohUNzFaREdJMERBVFhaWkIxUjFQRlpRRFJXMiQIQCN0PWcu>

The goal today is to complete a form that makes use of the information that we have been working on the last few "classes".

Please use the shortened periodic table to help and continue reading below this picture



The 1+ 2+ 3+ 4+/4- 3- 2- 1- 0 below the columns on the periodic table ARE THE CHARGE that the element will get once they have either lost or gained electrons (in order to be like the nearest noble gas). In other words, any element that is in the oxygen column will be a 2- (Se for example becomes a 2-)

Please use this to fill out this form. And, please fill out the form more than once if you need to. I want you to get perfect!

<https://forms.office.com/Pages/ResponsePage.aspx?id=GAMPRLReCU2WCd35yhGvQuASPjs6aYVfk-EAh60FvohUMVIVQVRTOUo1R1VKVUdHMzVFQzhIWihLNiQIQCN0PWcu>

(O) This message and/or attachment is intended for the sole use of the individual to which it is addressed and may contain information that is privileged and confidential. If the reader of this message is not the intended recipient or an authorized representative of the intended recipient, you are hereby notified that any dissemination of this communication is strictly prohibited. If you have received this communication in error, please notify me immediately and delete the message and any attachments from your system