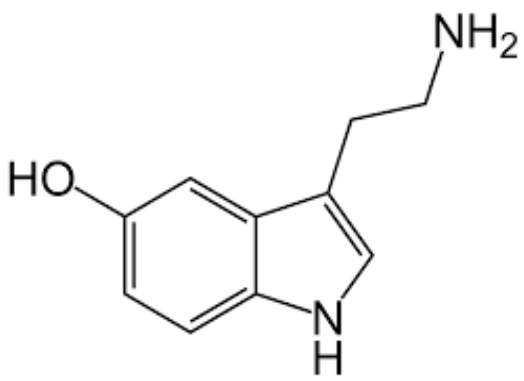
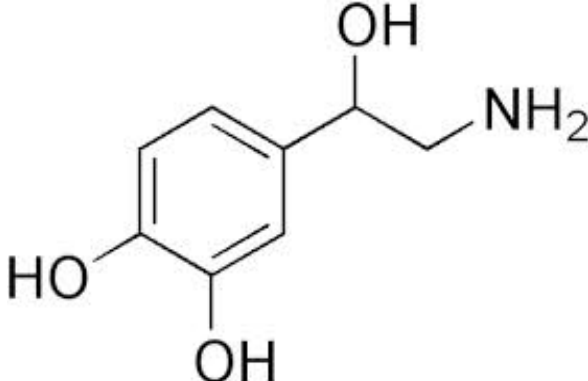


Name: _____

SCH 4U Organic Test - Part 1

1. State the degrees unsaturation and the chemical formula for serotonin and norepinephrine


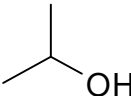
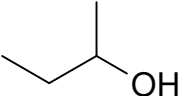
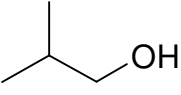
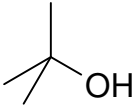
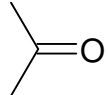
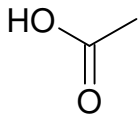
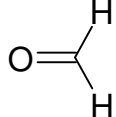

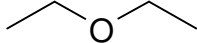

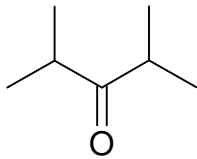
	
degree unsat = 6	degree unsat = 4
formula = C₁₀H₁₂N₂O	formula = C₈H₁₁NO₃

2. $H = [2C + 2] - 2(\text{deg. unsat}) - X + N$

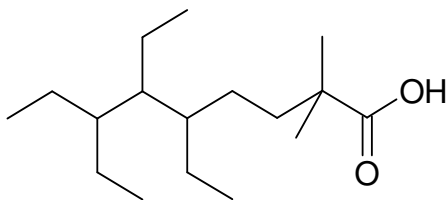
Use the above saturation formula to determine the degree of unsaturation and hence the possible combinations of functional groups (and rings) that would make possible each formula. (1/2 mark per correct response, some marks may be deduced for extra incorrect answers)

C ₁₂ H ₂₂ <ul style="list-style-type: none"> - alkene alkene - alkene ring - ring ring - alkyne 	C ₁₂ H ₂₄ O <ul style="list-style-type: none"> - aldehyde - ketone - alkene alcohol - alkene ether - ring alcohol - ring ether
C ₆ H ₁₃ NO <ul style="list-style-type: none"> - amide - aldehyde amine - ketone amine - alkene alcohol amine - alkene ether amine - ring alcohol amine - ring ether amine 	

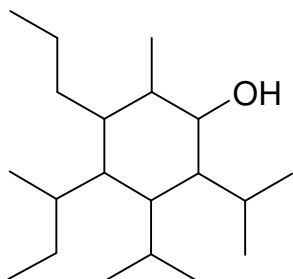
3. Provide common names and I.U.P.A.C. names for each of the following. If more than one common name exists, include both. Be sure to follow the rules when writing I.U.P.A.C. names. One mark per name

	Common Names	I.U.P.A.C.
	methyl alcohol	1-methanol
	isopropyl alcohol	2-propanol
	secbutyl alcohol	2-butanol
	isobutyl alcohol	2-methyl-1-propanol
	t-butyl alcohol	2-methyl-2-propanol
	dimethyl ketone acetone	2-propanone
	acetic acid	ethanoic acid
	formaldehyde	methanal
	acetylene	1-ethyne
	diethyl ether ether	
	diisopropyl ketone	2,4-dimethyl-3-pentanone

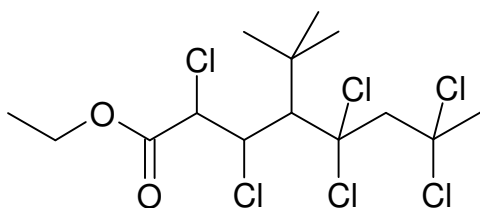
4. Provide full and correct I.U.P.A.C. names:



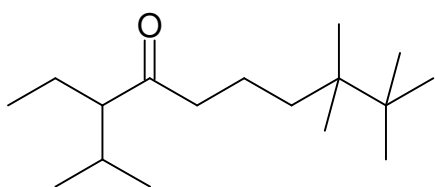
5,6,7-triethyl-2,2-dimethylnonanoic acid



4-secbutyl-2,3-diisopropyl-6-methyl-5-propyl-1-cyclohexanol



ethyl 4-t-butyl-2,3,5,5,7,7-hexachlorooctanoate



3-ethyl-2,8,8,9,9-pentamethyl-4-decanone

5. Provide all structural isomer for this formula. Be sure to consider unsaturation considerations that should be considered. I think that there are 22. Present your work in an organized fashion. Marks will be deducted for disorder. Also, marks will be deducted for duplicate (or triplicate etc. structures). Use only five and six member rings.

