

NOMENCLATURE #3 - SCH4C

Name: _____

Polyvalent Cations, Polyatomic Anions**Provide Names or Formula:**

1	NaClO_3	19	cobalt (II) nitrate
2	NaCl	20	lead (IV) sulphate
3	Na_2SO_4	21	sodium sulphate
4	Na_2S	22	tin (IV) chloride
5	$\text{Al}_2(\text{SO}_4)_3$	23	antimony (V) phosphate
6	KNO_3	24	aluminum sulphate
7	K_3PO_4	25	polonium (IV) chlorate
8	$\text{Au}_2(\text{CO}_3)_3$	26	iron (III) chlorate
		27	manganese (VII) nitrate
9	AgNO_3	28	gold (III) chlorate
10	KNO_3	29	silver carbonate
11	Li_3PO_4	30	gold (I) sulphate
12	NaClO_3	31	scandium phosphate
13	$\text{Al}(\text{ClO}_3)_3$	32	tin (IV) hydroxide
14	NaClO_3	33	hydrogen cyanide
15	FeCO_3	34	ammonium chloride
		35	calcium hydroxide
16	CuCO_3	36	potassium cyanide
		37	copper (II) phosphate
17	HgClO_3	38	ammonium carbonate
		39	tin (II) nitride
18	$\text{Pt}(\text{ClO}_3)_4$	40	tin (II) nitrate