

NOMENCLATURE #1 - SCH 4C

Name: _____

Monovalent Cations, Elemental Anions**Provide Names or Formula:**

1	MgCl ₂	magnesium chloride	21	potassium bromide	KBr
2	BaBr ₂	barium bromide	22	magnesium nitride	Mg ₃ N ₂
3	Rb ₂ S	rubidium sulphide	23	rubidium oxide	Rb ₂ O
4	CaF ₂	calcium fluoride	24	barium oxide	BaO
5	K ₂ S	potassium sulphide	25	lithium chloride	LiCl
6	Rb ₃ N	rubidium nitride	26	calcium iodide	CaI ₂
7	TlBr ₃	indium bromide	27	sodium telluride	Na ₂ Te
8	Al ₂ O ₃	aluminum oxide	28	sodium fluoride	NaF
9	SrS	strontium sulphide	29	barium phosphide	Ba ₃ P ₂
10	Mg ₃ N ₂	magnesium nitride	30	aluminum sulphide	Al ₂ S ₃
11	Ca ₂ C	calcium carbide	31	radium chloride	RaCl ₂
12	LiI	lithium iodide	32	strontium selenide	SrSe
13	H ₂ O	hydrogen oxide	33	francium oxide	Fr ₂ O
14	HCl	hydrogen chloride	34	hydrogen sulphide	H ₂ S
15	Be ₂ N ₃	beryllium nitride	35	magnesium arsenide	Mg ₃ As ₂
16	CaCl ₂	calcium chloride	36	beryllium iodide	BeI ₂
17	RaO	radium oxide	37	calcium oxide	CaO
18	KI	potassium iodide	38	potassium oxide	K ₂ O
19	GaAs	gallium(III) arsenide	39	potassium chloride	KCl
20	BaTe	barium telluride	40	zinc chloride	ZnCl ₂