

**NOMENCLATURE #4 - SCH 3U**  
**Polyvalent Cations, Polyatomic Anions**

Name: \_\_\_\_\_

**Provide Names or Formula:**

1	$\text{NaClO}_3$	19	cobaltous nitrate
2	$\text{NaCl}$	20	plumbic sulphate
3	$\text{Na}_2\text{SO}_4$	21	sodium sulphite
4	$\text{Na}_2\text{S}$	22	tin(IV) chloride
5	$\text{Al}_2(\text{SO}_4)_3$	23	antimonic phosphate
6	$\text{KNO}_3$	24	aluminum sulphate
7	$\text{K}_3\text{PO}_4$	25	polonium(IV) iodate
8	$\text{Au}_2(\text{CO}_3)_3$	26	ferric perchlorate
		27	manganese(VII) nitrite
9	$\text{AgNO}_3$	28	auric perchlorate
10	$\text{KNO}_2$	29	silver carbonate
11	$\text{Li}_3\text{PO}_3$	30	aurous sulphate
12	$\text{NaIO}_4$	31	scandium phosphate
13	$\text{Al}(\text{IO}_3)_3$	32	stannic hydroxide
14	$\text{NaBrO}_3$	33	hydrogen cyanide
15	$\text{FeCO}_3$	34	ammonium chloride
		35	calcium hydroxide
16	$\text{CuCO}_3$	36	potassium cyanide
		37	cupric phosphite
17	$\text{HgClO}_4$	38	ammonium carbonate
		39	stannous nitride
18	$\text{Pt}(\text{ClO}_4)_4$	40	stannous nitrate