

Cation	I.U.P.A.C. name	ous/ic name
C^{2+}	carbon(II)	carbonous
C^{4+}	carbon(IV)	carbonic
N^{3+}	nitrogen(III)	nitrous
N^{5+}	nitrogen(V)	nitric
P^{3+}	phosphorus(III)	phosphorous
P^{5+}	phosphorus(V)	phosphoric
S^{4+}	sulphur(IV)	sulphurous
S^{6+}	sulphur(VI)	sulphuric
Fe^{2+}	iron(II)	ferrous
Fe^{3+}	iron(III)	ferric
Co^{2+}	cobalt(II)	cobaltous
Co^{3+}	cobalt(III)	cobaltic
Ni^{2+}	nickel(II)	nickelous
Ni^{3+}	nickel(III)	nickelic
Cu^{1+}	copper(I)	cuprous
Cu^{2+}	copper(II)	cupric
As^{3+}	arsenic(III)	arsenous
As^{5+}	arsenic(V)	arsenic
Sn^{2+}	tin(II)	stannous
Sn^{4+}	tin(IV)	stannic
Sb^{3+}	antimony(III)	antimonous (stibous)
Sb^{5+}	antimony(V)	antimonic (stibic)
Au^{1+}	gold(I)	aurous
Au^{3+}	gold(III)	auric
Hg^{1+}	mercury(I)	mercurous
Hg^{2+}	mercury(II)	mercuric
Pb^{2+}	lead(II)	plumbous
Pb^{4+}	lead(IV)	plumbic
Mn^{2+}	manganese(II)	manganous
Mn^{4+}	manganese(IV)	manganic
Cr^{2+}	chromium(II)	chromous
Cr^{3+}	chromium(III)	chromic