

Name: \_\_\_\_\_

**Nomenclature #4 Mixed Questions**

Provide names or formula as appropriate:

1	lithium chloride		26	$\text{Ge}_3\text{N}_4$	
2	lithium sulphide		27	$\text{AgNO}_3$	
3	lithium sulphate		28	$\text{Ir}_3\text{N}_4$	
4	silver sulphate		29	$\text{Sn}_3(\text{PO}_4)_4$	
5	gold(III) sulphate		30	$\text{As}_2(\text{SO}_4)_3$	
6	gold(III) sulphide		31	$\text{K}_3\text{PO}_4$	
7	sodium hydroxide		32	$\text{Ca}(\text{OH})_2$	
8	potassium chlorate		33	$\text{SO}_3$	
9	calcium carbonate		34	$\text{Al}(\text{ClO}_3)_3$	
10	calcium carbide		35	$\text{FeO}$	
11	aluminum nitride		36	$\text{Fe}_2\text{O}_3$	
12	aluminum nitrate		37	$\text{CoCl}_2$	
13	silicon(IV) carbide		38	$\text{Sn}_3(\text{PO}_4)_2$	
14	lead(II) chloride		39	$\text{Mo}_2\text{O}_5$	
15	lead(II) sulphate		40	$\text{OsO}_4$	
16	copper(II) nitrate		41	$\text{Zr}(\text{CO}_3)_2$	
17	gold(I) carbonate		42	$\text{Cu}(\text{OH})_2$	
18	platinum(IV) oxide		43	$\text{Zn}(\text{NO}_3)_2$	
19	gallium(III) oxide		44	$\text{PbO}_2$	
20	zinc chlorate		45	$\text{CrO}_3$	
21	zinc chloride		46	$\text{W}(\text{SO}_4)_3$	
22	mercury(II) chloride		47	$\text{TlCl}$	
23	barium sulphate		48	$\text{UF}_6$	
24	arsenic(V) carbonate		49	$\text{Ce}_2(\text{SO}_4)_3$	
25	iodine(VII) chloride		50	$\text{Gd}(\text{NO}_3)_3$	