























Classifying Reactions and Balancing Chemical Equations


For each of the chemical reactions are listed below, complete the following:

 Balance the skeletal equation

 The type of chemical reaction (synthesis, decomposition, single displacement, or double displacement)


1.
 Balance the skeletal equation: $\underline{\quad} \text{Cu} + \underline{\quad} \text{O}_2 \rightarrow \underline{\quad} \text{CuO}$
 Reaction type: _____
2.
 Balance the skeletal equation: $\underline{\quad} \text{H}_2\text{O} \rightarrow \underline{\quad} \text{O}_2 + \underline{\quad} \text{H}_2$
 Reaction type: _____
3.
 Balance the skeletal equation: $\underline{\quad} \text{Fe} + \underline{\quad} \text{H}_2\text{O} \rightarrow \underline{\quad} \text{Fe}_2\text{O}_3 + \underline{\quad} \text{H}_2$
 Reaction type: _____
4.
 Balance the skeletal equation: $\underline{\quad} \text{H}_2\text{S} + \underline{\quad} \text{AsCl}_3 \rightarrow \underline{\quad} \text{As}_2\text{S}_3 + \underline{\quad} \text{HCl}$
 Reaction type: _____
5.
 Balance the skeletal equation: $\underline{\quad} \text{CaCO}_3 \rightarrow \underline{\quad} \text{CO}_2 + \underline{\quad} \text{CaO}$
 Reaction type: _____
6.
 Balance the skeletal equation: $\underline{\quad} \text{H}_2\text{S} + \underline{\quad} \text{KOH} \rightarrow \underline{\quad} \text{K}_2\text{S} + \underline{\quad} \text{HOH}$
 Reaction type: _____
7.
 Balance the skeletal equation: $\underline{\quad} \text{S}_8 + \underline{\quad} \text{Fe} \rightarrow \underline{\quad} \text{FeS}$
 Reaction type: _____
8.
 Balance the skeletal equation: $\underline{\quad} \text{H}_2\text{SO}_4 + \underline{\quad} \text{Al} \rightarrow \underline{\quad} \text{Al}_2(\text{SO}_4)_3 + \underline{\quad} \text{H}_2$
 Reaction type: _____
9.
 Balance the skeletal equation: $\underline{\quad} \text{H}_3\text{PO}_4 + \underline{\quad} \text{NH}_4\text{OH} \rightarrow \underline{\quad} (\text{NH}_4)_3\text{PO}_4 + \underline{\quad} \text{HOH}$
 Reaction type: _____
10.
 Balance the skeletal equation: $\underline{\quad} \text{O}_2 + \underline{\quad} \text{Al} \rightarrow \underline{\quad} \text{Al}_2\text{O}_3$
 Reaction type: _____

11.

 Balance the skeletal equation: $\underline{\quad} \text{H}_2\text{SO}_4 + \underline{\quad} \text{Al}(\text{OH})_3 \rightarrow \underline{\quad} \text{Al}_2(\text{SO}_4)_3 + \underline{\quad} \text{HOH}$

 Reaction type: _____

12.

 Balance the skeletal equation: $\underline{\quad} \text{Cl}_2 + \underline{\quad} \text{KBr} \rightarrow \underline{\quad} \text{KCl} + \underline{\quad} \text{Br}_2$

 Reaction type: _____

13.

 Balance the skeletal equation: $\underline{\quad} \text{Ca} + \underline{\quad} \text{HOH} \rightarrow \underline{\quad} \text{Ca}(\text{OH})_2 + \underline{\quad} \text{H}_2$

 Reaction type: _____

14.

 Balance the skeletal equation: $\underline{\quad} \text{H}_2\text{O}_2 \rightarrow \underline{\quad} \text{O}_2 + \underline{\quad} \text{H}_2\text{O}$

 Reaction type: _____

15.

 Balance the skeletal equation: $\underline{\quad} \text{Na} + \underline{\quad} \text{Cl}_2 \rightarrow \underline{\quad} \text{NaCl}$

 Reaction type: _____

16.

 Balance the skeletal equation: $\underline{\quad} \text{Zn} + \underline{\quad} \text{Pb}(\text{NO}_3)_2 \rightarrow \underline{\quad} \text{Zn}(\text{NO}_3)_2 + \underline{\quad} \text{Pb}$

 Reaction type: _____


17.

 Balance the skeletal equation: $\underline{\quad} \text{NaI} + \underline{\quad} \text{Pb}(\text{NO}_3)_2 \rightarrow \underline{\quad} \text{NaNO}_3 + \underline{\quad} \text{PbI}_2$

 Reaction type: _____


18.

 Balance the skeletal equation: $\underline{\quad} \text{P}_4 + \underline{\quad} \text{O}_2 \rightarrow \underline{\quad} \text{P}_2\text{O}_5$

 Reaction type: _____

19.

 Balance the skeletal equation: $\underline{\quad} \text{NH}_4\text{NO}_3 \rightarrow \underline{\quad} \text{H}_2\text{O} + \underline{\quad} \text{N}_2\text{O}$

 Reaction type: _____

20.

 Balance the skeletal equation: $\underline{\quad} \text{CaI}_2 + \underline{\quad} \text{AgNO}_3 \rightarrow \underline{\quad} \text{Ca}(\text{NO}_3)_2 + \underline{\quad} \text{AgI}$

 Reaction type: _____