

Name: \_\_\_\_\_

**Grade Ten Academic - Inorganic Nomenclature Quiz**

Provide names or formula as appropriate:

1	calcium chloride	
2	calcium sulphide	
3	calcium sulphate	
4	aluminum oxide	
5	aluminum carbonate	
6	iron(III) chloride	
7	iron(II) sulphide	
8	copper(I) oxide	
9	zinc hydroxide	
10	ammonium chloride	

11	NaCl	
12	Mg <sub>3</sub> N <sub>2</sub>	
13	Mg(NO <sub>3</sub> ) <sub>2</sub>	
14	Na <sub>3</sub> P	
15	Ba(OH) <sub>2</sub>	
16	Au <sub>2</sub> S <sub>3</sub>	
17	Au <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub>	
18	FeP	
19	Au <sub>2</sub> CO <sub>3</sub>	
20	HgCl <sub>2</sub>	

carbonate	CO <sub>3</sub> <sup>2-</sup>
nitrate	NO <sub>3</sub> <sup>1-</sup>
phosphate	PO <sub>4</sub> <sup>3-</sup>
sulphate	SO <sub>4</sub> <sup>2-</sup>
chlorate	ClO <sub>3</sub> <sup>1-</sup>
hydroxide	OH <sup>1-</sup>
cyanide	CN <sup>1-</sup>
ammonium	NH <sub>4</sub> <sup>1+</sup>

carbide  
nitride  
oxide  
fluoride  
phosphide  
sulphide  
chloride  
arsenide  
selenide  
bromide  
antimonide  
telluride  
iodide