

Name: \_\_\_\_\_

**Grade Ten Academic - Inorganic Nomenclature Quiz**

Provide names or formula as appropriate:

1	calcium chloride	$\text{CaCl}_2$
2	calcium sulphide	$\text{CaS}$
3	calcium sulphate	$\text{CaSO}_4$
4	aluminum oxide	$\text{Al}_2\text{O}_3$
5	aluminum carbonate	$\text{Al}_2(\text{CO}_3)_3$
6	iron(III) chloride	$\text{FeCl}_3$
7	iron(II) sulphide	$\text{FeS}$
8	copper(I) oxide	$\text{Cu}_2\text{O}$
9	zinc hydroxide	$\text{Zn}(\text{OH})_2$
10	ammonium chloride	$\text{NH}_4\text{Cl}$

11	$\text{NaCl}$	sodium chloride
12	$\text{Mg}_3\text{N}_2$	magnesium nitride
13	$\text{Mg}(\text{NO}_3)_2$	magnesium nitrate
14	$\text{Na}_3\text{P}$	sodium phosphide
15	$\text{Ba}(\text{OH})_2$	barium hydroxide
16	$\text{Au}_2\text{S}_3$	gold(III) sulphide
17	$\text{Au}_2(\text{SO}_4)_3$	gold(III) sulphate
18	$\text{FeP}$	iron(III) phosphide
19	$\text{Au}_2\text{CO}_3$	gold(I) carbonate
20	$\text{HgCl}_2$	mercury(II) chloride

carbonate	$\text{CO}_3^{2-}$
nitrate	$\text{NO}_3^{1-}$
phosphate	$\text{PO}_4^{3-}$
sulphate	$\text{SO}_4^{2-}$
chlorate	$\text{ClO}_3^{1-}$
hydroxide	$\text{OH}^{1-}$
cyanide	$\text{CN}^{1-}$
ammonium	$\text{NH}_4^{1+}$

carbide
nitride
oxide
fluoride
phosphide
sulphide
chloride
arsenide
selenide
bromide
antimonide
telluride
iodide