

Name: _____

Atomic Symbols

<u>scandium</u>	45 Sc 21	# of p ⁺ = <u>21</u> # of n = <u>24</u> # of e ⁻ = <u>21</u>
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<u>magnesium</u>	24 Mg 12	# of p ⁺ = <u>12</u> # of n = <u>12</u> # of e ⁻ = <u>12</u>
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<u>potassium</u>	39 K 19	# of p ⁺ = <u>19</u> # of n = <u>20</u> # of e ⁻ = <u>19</u>
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<u>uranium</u>	238 U 92	# of p ⁺ = <u>92</u> # of n = <u>146</u> # of e ⁻ = <u>92</u>
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<u>gold</u>	197 Au 79	# of p ⁺ = <u>79</u> # of n = <u>118</u> # of e ⁻ = <u>79</u>
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<u>iron</u>	56 Fe 26	# of p ⁺ = <u>26</u> # of n = <u>30</u> # of e ⁻ = <u>26</u>
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<u>chlorine</u>	35 Cl 17	# of p ⁺ = <u>17</u> # of n = <u>18</u> # of e ⁻ = <u>17</u>
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<u>bromine</u>	80 Br 35	# of p ⁺ = <u>35</u> # of n = <u>45</u> # of e ⁻ = <u>35</u>
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<u>arsenic</u> <u>75</u> <u>33</u>	<u>As</u>	# of p ⁺ = <u>33</u> # of n = <u>42</u> # of e ⁻ = <u>33</u>
<u>silver</u> <u>109</u> <u>47</u>	<u>Ag</u>	# of p ⁺ = <u>47</u> # of n = <u>62</u> # of e ⁻ = <u>47</u>
<u>sodium</u> <u>23</u> <u>11</u>	<u>Na</u>	# of p ⁺ = <u>11</u> # of n = <u>12</u> # of e ⁻ = <u>11</u>
<u>lead</u> <u>207</u> <u>82</u>	<u>Pb</u>	# of p ⁺ = <u>82</u> # of n = <u>125</u> # of e ⁻ = <u>82</u>
<u>platinum</u> <u>195</u> <u>78</u>	<u>Pt</u>	# of p ⁺ = <u>78</u> # of n = <u>117</u> # of e ⁻ = <u>78</u>
<u>copper</u> <u>63</u> <u>29</u>	<u>Cu</u>	# of p ⁺ = <u>29</u> # of n = <u>34</u> # of e ⁻ = <u>29</u>
<u>chromium</u> <u>52</u> <u>24</u>	<u>Cr</u>	# of p ⁺ = <u>24</u> # of n = <u>28</u> # of e ⁻ = <u>24</u>
<u>iodine</u> <u>127</u> <u>53</u>	<u>I</u>	# of p ⁺ = <u>53</u> # of n = <u>74</u> # of e ⁻ = <u>53</u>